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A New C–C Bond-forming Free Radical Rearrangement

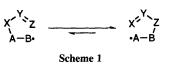
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On treatment with tributyltin hydride and azoisobutyronitrile in benzene at reflux α -aryl- β -bromoalkyl vinyl ethers rearrange in moderate yield to δ -aryl ketones: possible mechanistic rationales are discussed for this novel C–C bond-forming process.

The formalism of Scheme 1 conveniently describes a number of free radical 2,3-shifts. These are the well known acyloxy migration (A = B = Y = carbon, X = Z = oxygen)¹ and the allylperoxy migration (A = B = oxygen, X = Y = Z = carbon)² together with the more recent phosphonooxy migration [A = B = carbon, X = Z = oxygen, Y = phosphorus(v)]³ and a number of other related migrations.^{4,5}

Each of these rearrangements, which involve the making and breaking of carbon-heteroatom bonds, is of considerable mechanistic interest but has only limited synthetic utility.^{6,7} We were intrigued by, and report here on the possibility of using Scheme 1 as a framework for the design of related



migrations that would occur with formation of a C–C bond (*i.e.* in which at least B = Z = carbon) and so a synthetically useful process.

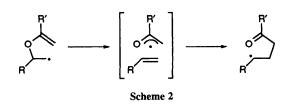
The concept was readily put into practice. Reaction of the Tebbe reagent⁸ with styrene bromohydrin acetate gave the enol ether 1 which on treatment in benzene at reflux with

Entry	Substrate	Products (% Yield) ^a
1		$Ph(CH_{2})_{3}COMe$ 2^{17} 32 (50) + $Ph \downarrow O$ 3 (50)
2	Br Ph 4	$Ph(CH_2)_3COEt$ 5 ¹⁷ 28 (53) + $Ph \downarrow_O$ 6 (47)
3	Br ₽₩00-CgH₄ O Ph	$\begin{array}{cccc} p \text{MeO-C}_{6}\text{H}_{4}(\text{CH}_{2})_{3}\text{COPh} & + & PhCOMe \\ \hline & & & & \\ 8 & 39 & (50) & & & \\ & & & & \\ p \text{MeO-C}_{6}\text{H}_{4} & & & \\ & & & & \\ \hline & & & & \\ 9^{b} & 19.5 & (28) & & \\ \end{array} + & & & \\ & & & & \\ P \text{MeO-C}_{6}\text{H}_{4}\text{CH} = \text{CH}_{2} \\ \hline & & & \\ \end{array}$
4	^{Вг} р-МеО-С ₆ Н ₄ -р-Г 10	$PM_{\Theta}O-C_{\theta}H_{4}(CH_{2})_{3}COC_{\theta}H_{4}\cdot PF + PF-C_{\theta}H_{4}COM_{\theta}$ 11 38 (54) (23)
5	0	$\bigcup_{14 \ 9 \ (15)}^{\circ} + \bigcup_{15 \ (85)}^{\circ}$
6	Br Ph O	PhCH=CH ₂ (>95)
7	Ph 17	Ph 18 (>95)

^a Yields refer to isolated products wherever possible. ¹H NMR yields determined on the crude reaction mixtures after removal of volatiles under vacuum, by comparison with authentic samples, are given in parentheses. ^b cis/trans = 1.7/1. ^c cis/trans = 2.0/1. Z/E = 3.5/1.

tributyltin hydride (TBTH) and azoisobutyronitrile (AIBN) over 18 h resulted in the formation of the migration product 2 together with the direct reduction product 3 (Table, entry 1). A number of other examples, prepared similarly by application of the Tebbe or Oshima-Takai9 protocols to the corresponding esters are presented in Table 1 and serve to define the present scope of the rearrangement. The major byproducts are either the straight reduction products, as in entries 1 and 2, or in the case of the 1-aryl-1-alkoxyethenes 7 and 10 the corresponding substituted acetophenones and tetrahydrofurans (Table 1, entries 3 and 4). Substitution at either end of the purported new C-C bond (Table 1, entries 5 and 6) results in drastically reduced yields and the isolation of alkenes. Finally, migration was not observed for the enol ether derivative 17 of a simple aliphatic bromohydrin derived from camphor (Table 1, entry 7).

On the basis of the above observations we feel justified in considering a mechanism involving radical elimination followed by cage recombination (Scheme 2).[†] The elimination is



accelerated by formation of a conjugated alkene, and hence the minimization of the reduction product in all cases but one (Table 1, entry 7). Readdition is retarded by substitution at the addition site, a hallmark of radical addition reactions.¹⁰

However, an alternative pathway involving ring closure with formation of an α -tetrahydrofuranyl radical followed by ring opening cannot be ruled out although it is disfavoured by Baldwin's rules,¹¹ and is somewhat unlikely in view of the known reluctance of such radicals to suffer ring opening¹² except under drastic conditions¹³ as well as of their ready participation in preparatively useful chemistry.¹⁴ The formation of tetrahydrofurans 9 and 12¹⁵ as byproducts from 7 and 10, respectively, is noteworthy and indicates that 5-*endo-trig* cyclization at least occurs as a side rection when benzylic stabilization is available to the ring closed radical.¹⁶ Efforts are currently underway to extend the scope of this new C–C

[†] A crossover experiment involving heating 7 and TBTH in the presence of styrene gave no indication of the formation of the crossover product pointing to an intramolecular mechanism.

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bond forming rearrangement and will be reported in due course.‡

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 $[\]ddagger$ All new compounds gave spectal data and, with the exception of enol ethers **3** and **6** which could not be isolated without decomposition, HRMS/microanalytical data in accordance with the assigned structures.